

L03 EFFECT OF HEXAMETHYLDISILOXANE SOLUTION OF ZIRCONIUM PROPOXIDE ON PERMANENCE OF PAPER

SOŇA KIRSCHNEROVÁ, KATARÍNA VIZÁROVÁ and SVETOZÁR KATUŠČÁK

*Institute of Polymer Materials, Department of Chemical Technology of Wood, Pulp and Paper, Faculty of Chemical and Food Technology, Slovak University of Technology, Radlinského 9, 831 07 Bratislava, Slovakia
sona.kirschnerova@stuba.sk*

Introduction

Ageing of lignocellulosic carries of information is natural process influenced by different factors. Of the outdoor impacts, relative humidity, oxides of sulphur and nitrogen and others, while of the indoor factors, type of fibres, sizing agent, filling agent, etc. may be introduced^{1,2,3}.

Influence of ageing leads to formation of other compounds, which accelerate degradation of paper, e.g. organic compounds^{4,5,6}.

To stop occurring these processes, conservation techniques, such as deacidification of paper are used⁷. Efficient deacidification systems have been applied in both polar and nonpolar medium in conservation practice. If this modification system is in the polar medium, it is supposed that diffusion of low molecular compounds to the cell wall, fibrils, microfibrils, elementary fibrils and to the molecular level of other units are to be performed.

Double alkoxides, combination of chemical elements Al-K, Sn-Mg, Zr-Ca are used as a neutralizing agent in suitable solvents, such as alcohols, fluoro-chlorohydrocarbons, hydrocarbons and siloxanes. Disadvantage of the above mentioned alkoxides lies in the formation of visible deposit^{8,9}.

For papermaking industry, water solutions of zirconium compounds are of particular importance, for example adhesions supporting cohesion to the different surface, etc¹⁰.

This work was aimed at verifying the effect of hexamethyldisiloxane (HMDO) solution of zirconium propoxide on the permanence of newsprint paper.

Experimental

Raw Materials

Newsprint paper with extract pH 5.5–6.0; grammage $m_s = 45 \text{ g m}^{-2}$, containing 55 % of mechanical bleached groundwood, 20 % of bleached kraft pulp, 15 % scrap fibres and 10 % clay was used as received.

Modification System

Two different solution of zirconium propoxide (70 % wt. in 1-propanol, Aldrich) in hexamethyldisiloxane ($\geq 98 \%$, Aldrich): 0.02M solution and 0.04M solution were used.

Sheets were modified for 10 minutes at regulat stirring in a laboratory shaker. Subsequently, sheets were dried free in sieves.

Accelerated Ageing

Samples of paper were conditioned for 24 hours at $T = (23 \pm 1) \text{ }^\circ\text{C}$, $\text{RH} = (50 \pm 2) \%$ before the accelerated ageing. Subsequently were aged in hermetically sealed bags at $98 \pm 2 \text{ }^\circ\text{C}$ during 0, 1, 2, 3 and 5 days according to standard ASTM D 6819-02¹¹.

Property Measurement

Changes of folding endurance¹² (ω), breaking length¹³ (lt), Kubelke-Munke intensity of $1,735 \text{ cm}^{-1}$ peak by FTIR¹⁴ (KM intensity), $\text{pH}_{\text{extract}}$ ¹⁵ and total color difference¹⁶ (ΔE) were studied at modified and aged samples.

Folding endurance was expressed as a coefficient of permanence^{17, 18} in the time of ageing (t):

$$S_{\omega,t} = \frac{\omega_{\text{modified},t}}{\omega_{\text{control},t}} \quad (1)$$

- if $S_{\omega,t} > 1$ – permanence increased
- if $S_{\omega,t} < 1$ – permanence decreased
- if $S_{\omega,t} = 1$ – is it not change.

Results

The permanence of paper samples modified with HMDO solution of zirconium propoxide was studied. Permanence of paper was evaluated by the change of folding endurance (ω), breaking length (lt), KM intensity of $1,735 \text{ cm}^{-1}$ peak (KM intensity), $\text{pH}_{\text{extract}}$, and total color difference (ΔE).

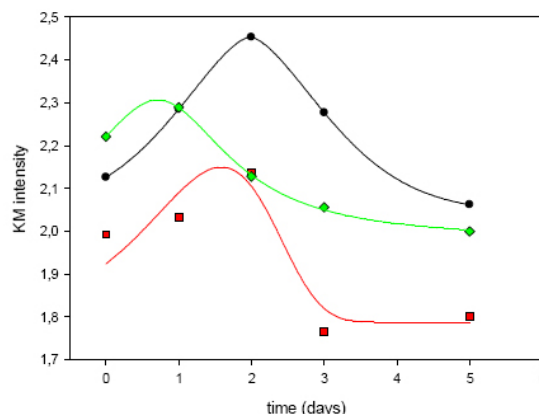


Fig. 1. Dependence of KM intensity peak $1,735 \text{ cm}^{-1}$ on the time of accelerated ageing obtained by evaluation of FTIR DRIFT spectra of control (●) and samples modified by 0.02M zirconium propoxide in HMDO (■) and 0.04M zirconium propoxide in HMDO (◆)

Increased formation of carbonyls and carboxyls depends on time of ageing. In Fig. 1., values of KM intensity absorption peak at $1,735 \text{ cm}^{-1}$, which relate to binding vibration C=O groups are shown. Modification system restricts creation of carbonyls and carboxyls. Marked decrease of absorption peak at $1,735 \text{ cm}^{-1}$ was observed at lower concentration of effective component compared with the control sample during the modification. This trend is held during all

accelerated ageings. Higher concentration of zirconium propoxide in HMDO causes increased formation of carbonyls and carboxyls with unaged control sample immediately after the modification. However during the accelerated ageing a decrease compared with control sample is observed.

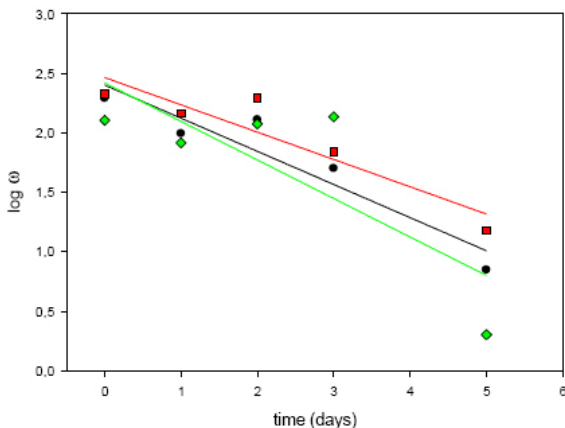


Fig. 2. Dependence of common logarithm values of folding endurance during the accelerated ageing for samples modified by 0.02M zirconium propoxide in HMDO (■), 0.04M zirconium propoxide in HMDO (◆) and control sample (●)

Folding endurance moderately increased at lower concentration of effective component ($S_{\omega,0} = 1.1$) just after the modification. During the accelerated ageing, stability values of folding endurance compared with unmodified control sample were followed. Stability effect of folding endurance after 5 days of accelerated ageing reached value 2.1 ($S_{\omega,5} = 2.1$). Higher concentration of zirconium propoxide in HMDO influences only slightly the folding endurance. It leads to an expressive decrease of this property after 0 and 5 days ($S_{\omega,0} = 0.7$ a $S_{\omega,5} = 0.3$).

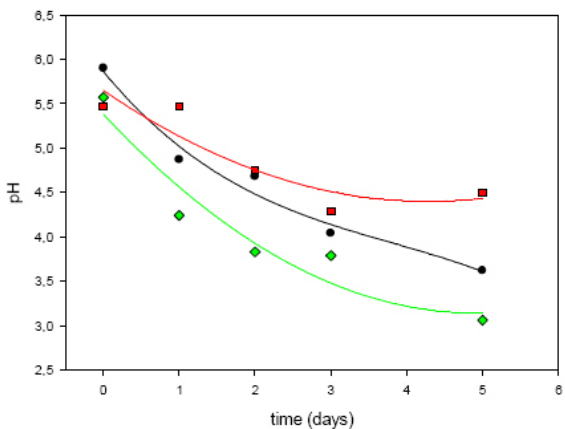


Fig. 3: Dependence of pH of cold extract samples modified by 0.02M zirconium propoxide in HMDO (■), 0.04M zirconium propoxide in HMDO (◆) and control – unmodified sample (●) during the accelerated ageing.

Cold extract of pH modified samples are comparable with unmodified sample ($\text{pH} \approx 5.5$). However pH of sample with lower concentration of zirconium propoxide in HMDO ($\text{pH} = 4.5$) is more mild decreased after 5 days of accelerated ageing in comparison with unmodified control sample ($\text{pH} = 3.6$).

Table I

Breaking length values of samples modified with 0.02M zirconium propoxide in HMDO (0.02M), 0.04M zirconium propoxide in HMDO (0.04M) and unmodified samples (control) during accelerated ageing

Days	Control	0.02 M	0.04M
0	3,720 ± 570	3,860 ± 280	3,830 ± 410
1	3,760 ± 490	3,790 ± 390	4,000 ± 230
2	3,280 ± 670	3,770 ± 310	3,530 ± 550
3	3,600 ± 380	3,460 ± 490	3,560 ± 270
5	2,910 ± 430	3,340 ± 320	3,200 ± 380

Values of breaking length obtained from modified samples were higher than those of unmodified ones.

Samples modified with zirconium propoxide in hexamethyldisiloxane show decrease of measured properties in comparison with unmodified control sample. System of modification suppresses the formation of carbonyls and carboxyls. It indicates retardation of ageing process. Effect of modification causes that brittleness of paper decreases and that folding endurance rises. Disadvantage of this modification system is low values of pH, which are comparable with unmodified paper. Combination of studied system with compound able to increase pH and thus alkaline reserve (for an instance MgO or other from lines of magnesium or calcium compounds) can help to solve this problem.

Advantage of this modification system is in the presence of polar element, 1-propanol, which is solvent of zirconium propoxide and at the same time, offers the use of non-polar carrier of effective compound. Non-polar solvent causes smooth surface of modified paper without warps and formation of visible deposits on the surface. Polar ingredient compiles presumption of facilitate intersection of effective zirconium compound to the structure of paper.

Lower concentration is more favourable in term of observed properties. This fact is shown on the look of modified paper. At a higher concentration of the effective component, e.g., at 0.04M solution, the total colour difference measured vs. the original unmodified paper is much more expressive ($\Delta E = 23.6$) than that at lower concentration of zirconium compound ($\Delta E = 17.5$).

Conclusions

Solution of zirconium propoxide in HMDO exhibits a positive influence on the permanence of paper, and/or on measured properties. Utilization of studied modification system resulted in smooth surface of modified paper without warps and formation of visible deposits on the surface.

It leads to a decreased formation of carbonyl and carboxyl groups. There was also observed decrease intensity of the peak at $1,735\text{ cm}^{-1}$ (carbonyl groups) in comparison with control sample measured by FTIR.

It enhances the effect of stability of folding endurance – decreases brittleness of the measured sample in comparison with unmodified one.

Artificial ageing causes decrease of l_t by about 15 % in comparison with the control paper.

A more favourable effect offered solution of zirconium propoxide with lower concentration. Disadvantage of this system is its pH in comparison with pH of unmodified paper.

Combination of studied system with compound able to increase pH and thus alkaline reserve (for an instance MgO or other from lines of magnesium or calcium compounds) can help to solve this problem.

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